Magnesium Salicylate 467.2 mg Tablet

Structure:

Molecular Formula and Mass: C₁₄H₁₀MgO₆ – 298.533

Category: Analgesic – Nonsteroidal Anti-inflammatory (NSAID)

Sample:

Grind one tablet and dissolve in 100 mL of ethanol. Shake at least 10 min and filter. 467 mg/100 mL = 4.67 mg/mL. Further dilute 1.00 mL with an additional 2.00 mL of ethanol, for a total volume of 3.00 mL. Final concentration of sample solution = 4.67 mg/3.00 mL = 1.56 mg/mL, which is the required concentration representing 100%.

Standards:

High Standard:

The high limit is 115%; therefore the concentration of the high standard = (1.56 mg/mL X 1.15 = 1.79 mg/mL. Weigh approximately 17.9 mg of standard. If you weighed 18.0 mg of standard, dissolve it in: (18.0 mg)/(1.79 mg/mL) = 10.1 mL of ethanol. This makes the high standard solution concentration equal to 1.79 mg/mL. [If magnesium salicylate standard is not available, salicylic acid standard can be used. Weigh approximately 16.6 mg of salicylic acid standard. If you weighed 16.7 mg of standard, dissolve it in: (16.7 mg)/(1.66 mg/mL) = 10.1 mL of ethanol. This makes the high standard solution concentration to 1.66 mg salicylic acid/mL, which is equivalent to 1.79 mg magnesium salicylate/mL.]

Low Standard:

The low limit is 85%; therefore the concentration of the low standard = (1.56 mg/mL X 0.85 = 1.33 mg/mL. Dilute 1.00 mL of high standard to 1.35 mL by adding 0.35 mL of methanol (1.15/0.85 = 1.35).

Spotting:

Spot on the 5 X 10 cm silica gel TLC aluminium plate with 3.00 μL aliquots as follows:

Left spot low standard (85%) = $3.98 \mu g$

Center Spot 100% sample = $4.68 \mu g$

Right Spot high standard (115%) = $5.38 \mu g$

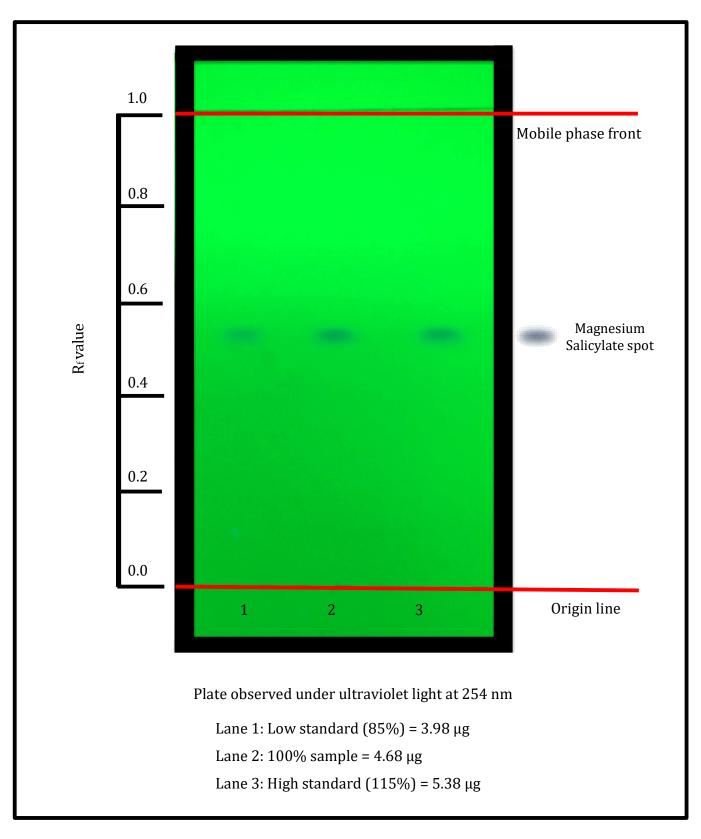
Development:

Mix 95.0 mL of ethyl acetate and 5.00 mL of glacial acetic acid. Develop the plate in a small glass chamber with approximately 20.0 mL of this solution until the solvent front reaches within 1 cm of the top of the TLC plate. Both magnesium salicylate and salicylic acid migrate with the same $R_f = 0.56$.

Detection:

UV:

Dry the plate and observe under ultraviolet light at 254 nm. Observe the intensities and the sizes of the spots.



Developed and tested by Kaitlin Nguyen and Joseph Sherma Department of Chemistry, Lafayette College, Easton, PA, USA July 2017

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